

In-situ nanoscale characterization of cyclic reduction and reoxidation of CeO_x islands on Cu(111)

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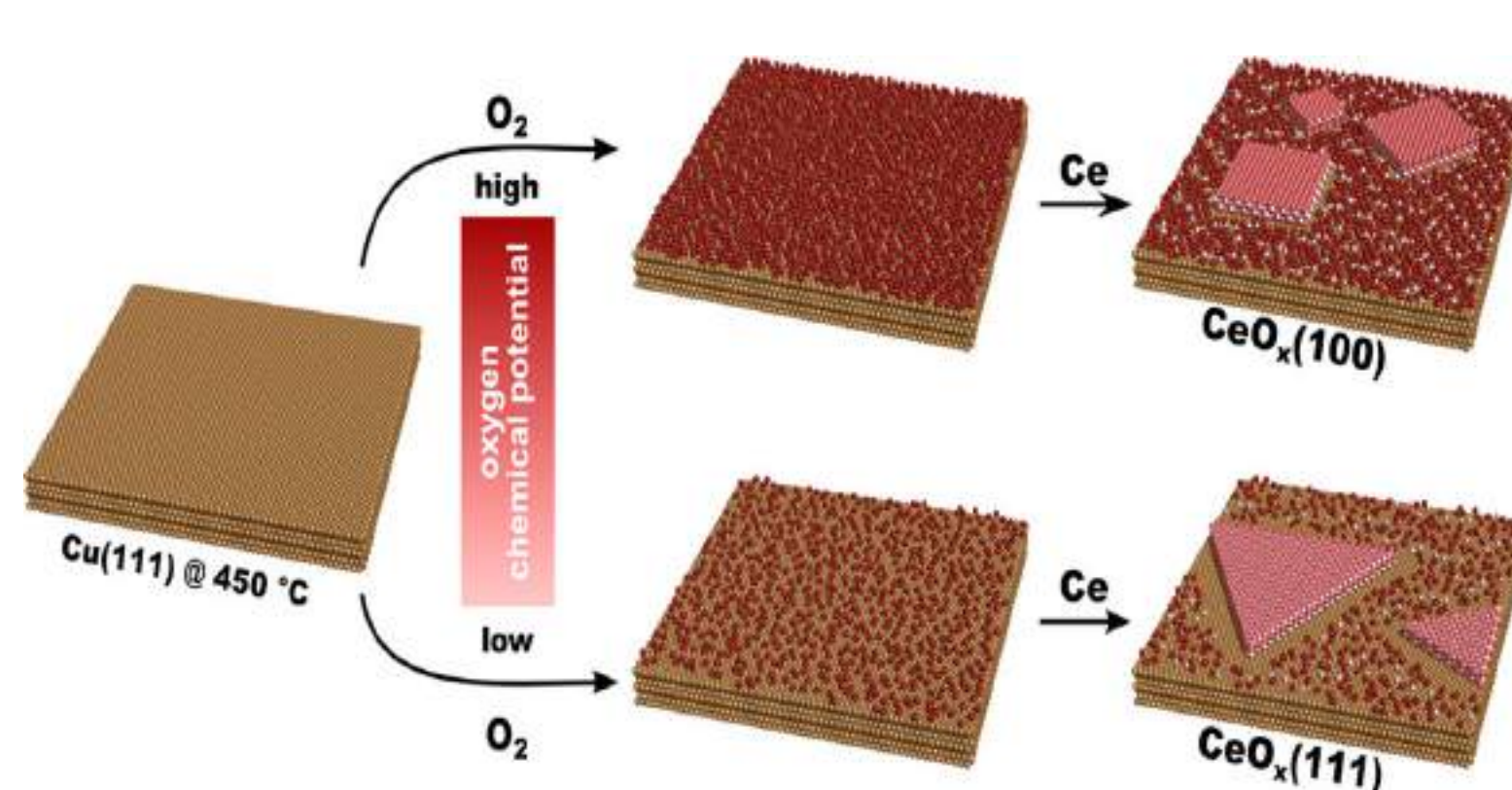
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Introduction

- Methanol is a prototypical synthetic fuel.
- CeO_x/Cu(111) is a highly active catalyst for CO₂ conversion to methanol (left) [1].
- CeO_x(100) islands may also be active [2]
- controlled side-by-side growth [3].
- H₂ exposure leads to Ce³⁺ sites, which can activate CO₂ molecules and are essential for methanol synthesis [1,4].



Experimental

- 1. Ceria island growth:**
 - Substrate: Cu(111)
 - $T = 470\text{ °C}$, $p(\text{O}_2) = 2 \cdot 10^{-6}\text{ mbar}$
 - evaporation of metallic Ce
- 2. Treatments (550 °C)**
 - Reduction: $p(\text{H}_2) \leq 1 \cdot 10^{-5}\text{ mbar}$, H₂ dose: 179 kL
 - Reoxidation: $p(\text{CO}_2) \leq 1 \cdot 10^{-4}\text{ mbar}$, CO₂ dose: 145 kL
 - 2nd redox cycle: $p(\text{H}_2, \text{CO}_2) = 1 \cdot 10^{-5}\text{ mbar}$, H₂^{max} dose: 14.4 kL, CO₂^{max} dose: 28.4 kL

In situ growth of CeO₂/Cu(111)

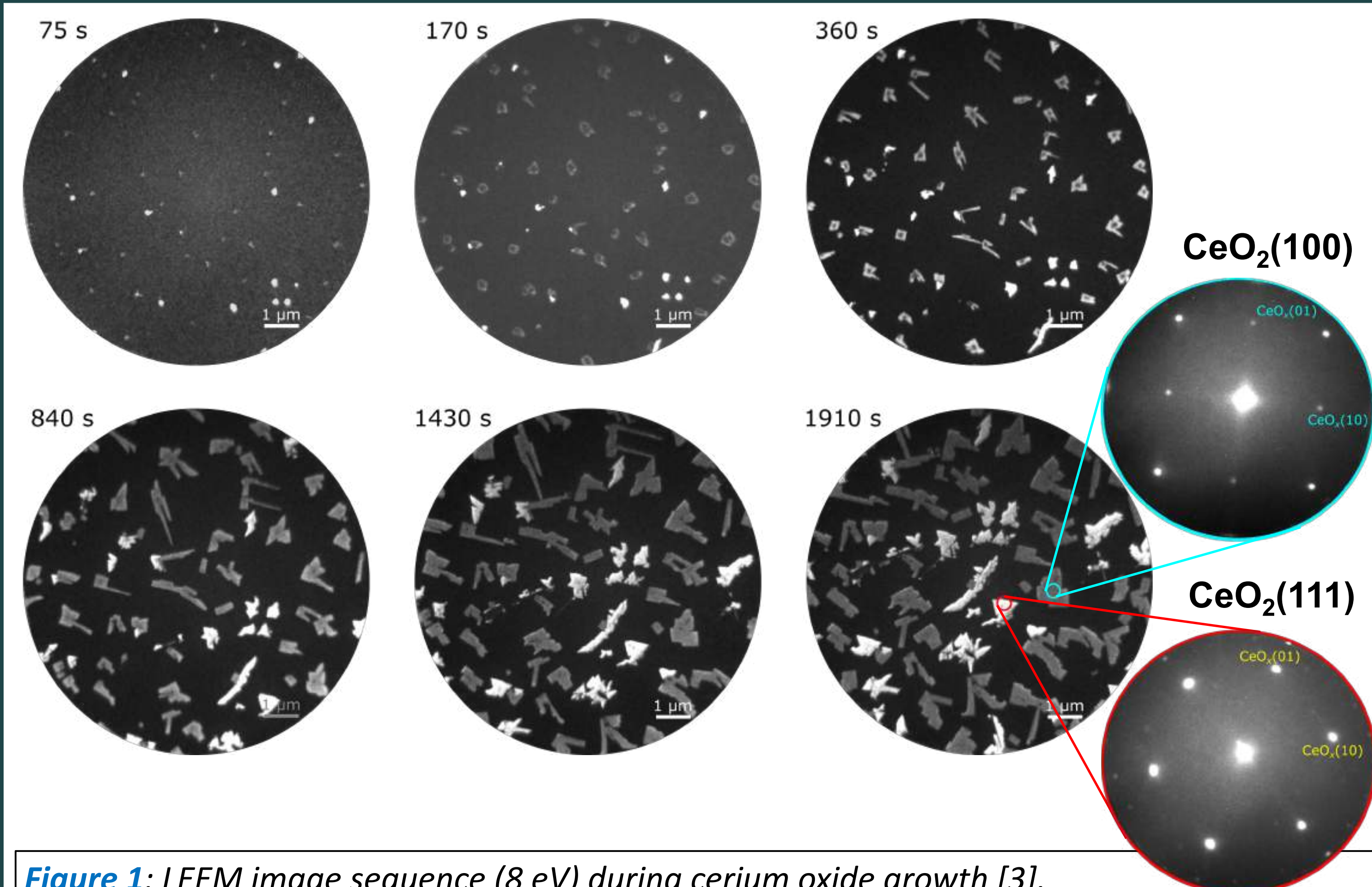


Figure 1: LEEM image sequence (8 eV) during cerium oxide growth [3]. LEED images taken at 200 °C (25 eV) after growth.

Figure 2: XAS-PEEM spectra for reduction with H₂ (left, from [5]), during reduction with H₂ (middle), and during reoxidation with CO₂ (right).

Figure 3: I(V)-LEEM curves during reduction with H₂ (left) and reoxidation with CO₂ (right).

Reduction and reoxidation: XAS-PEEM @ Ce M_{4,5} edge

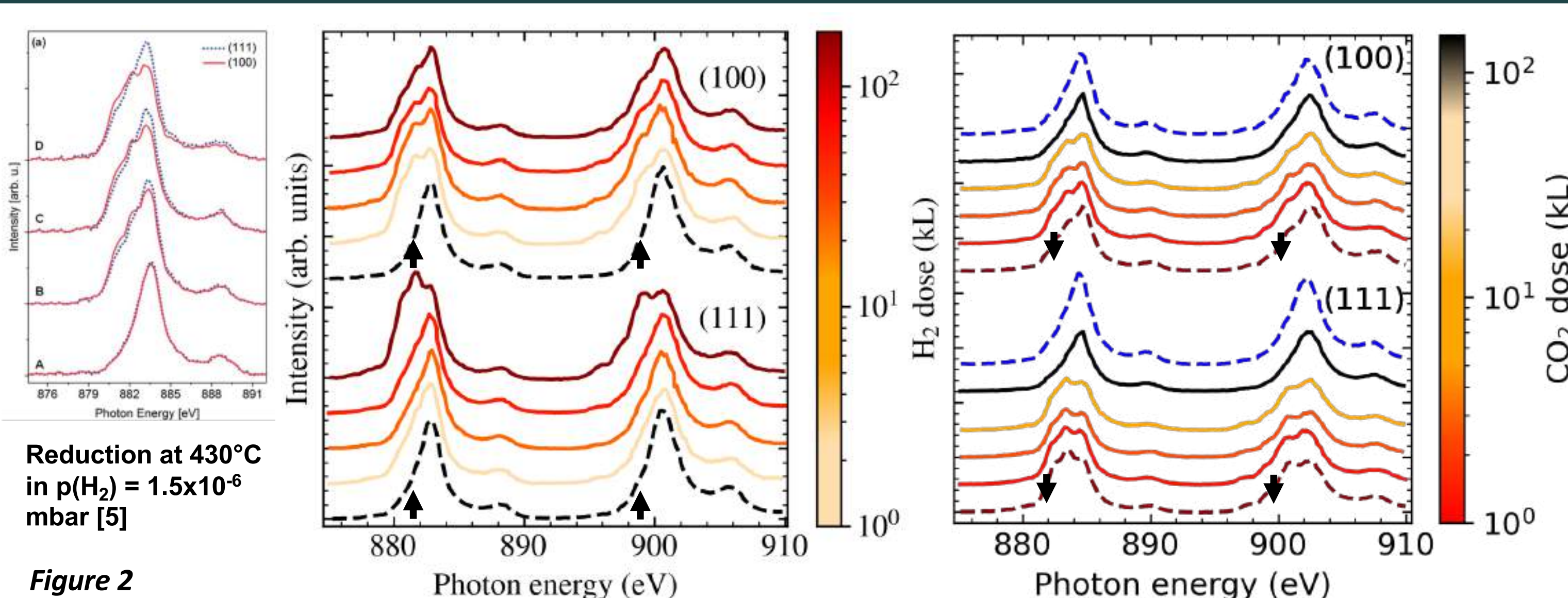


Figure 2

Reduction and reoxidation: I(V)-LEEM of single islands

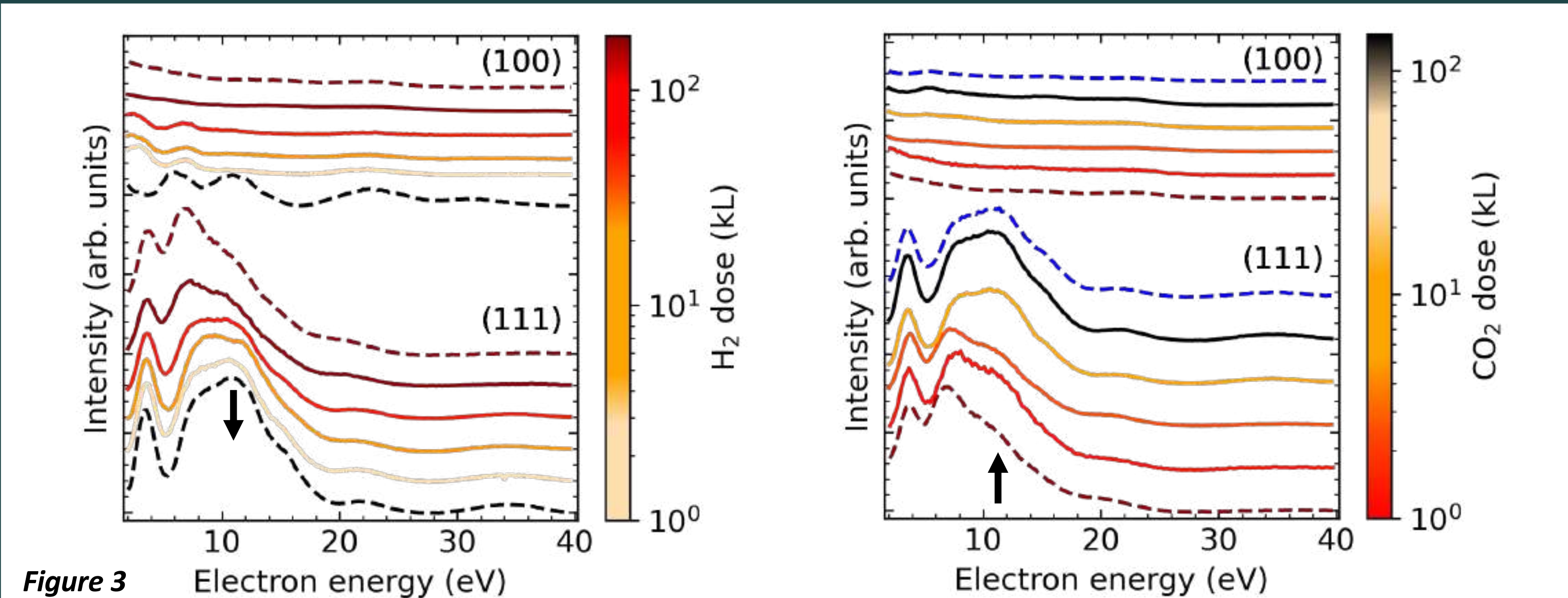


Figure 3

Dynamic I(V)-LEEM with pixel-wise analysis: 2nd redox cycle of CeO₂(111)

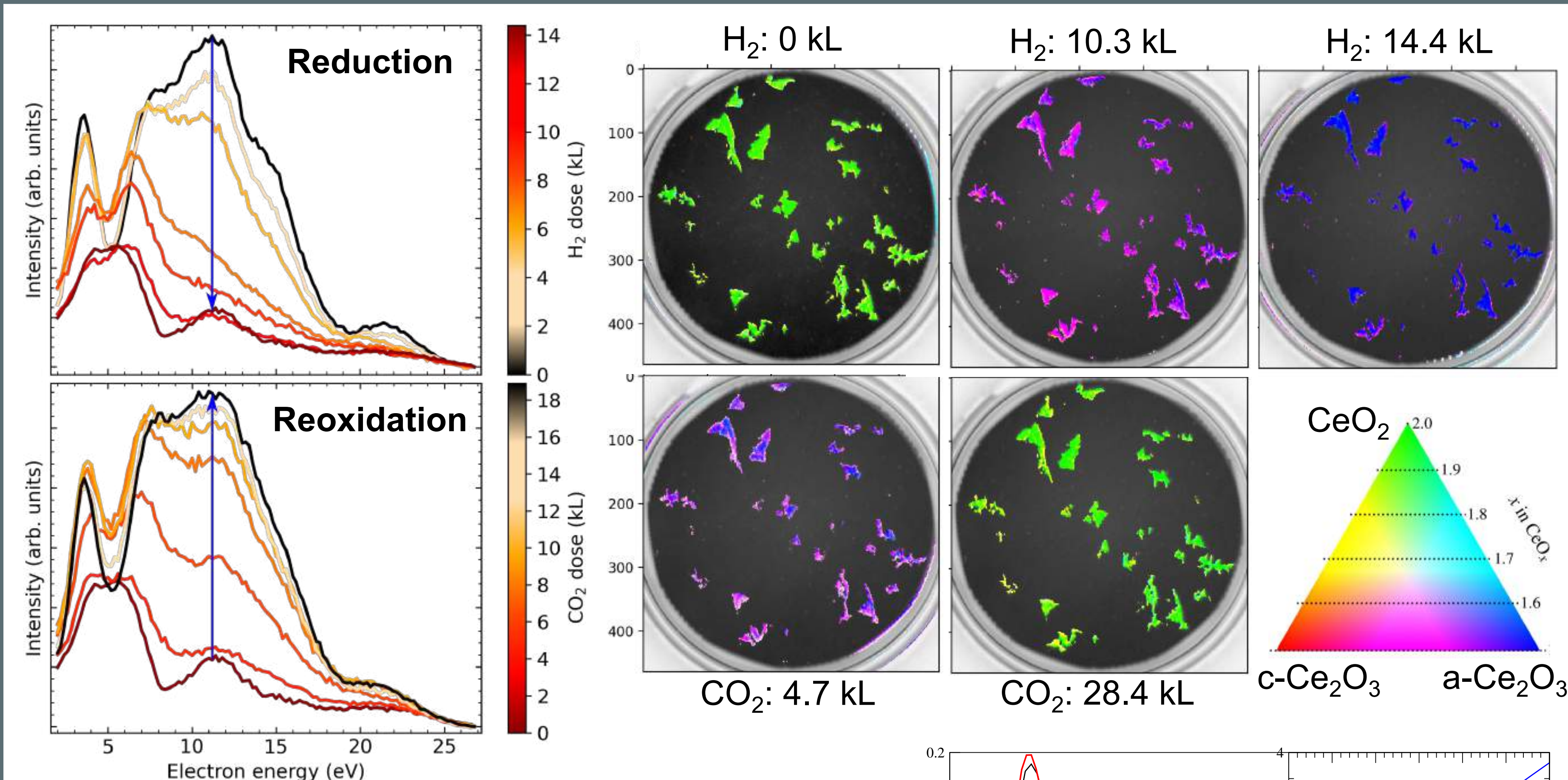
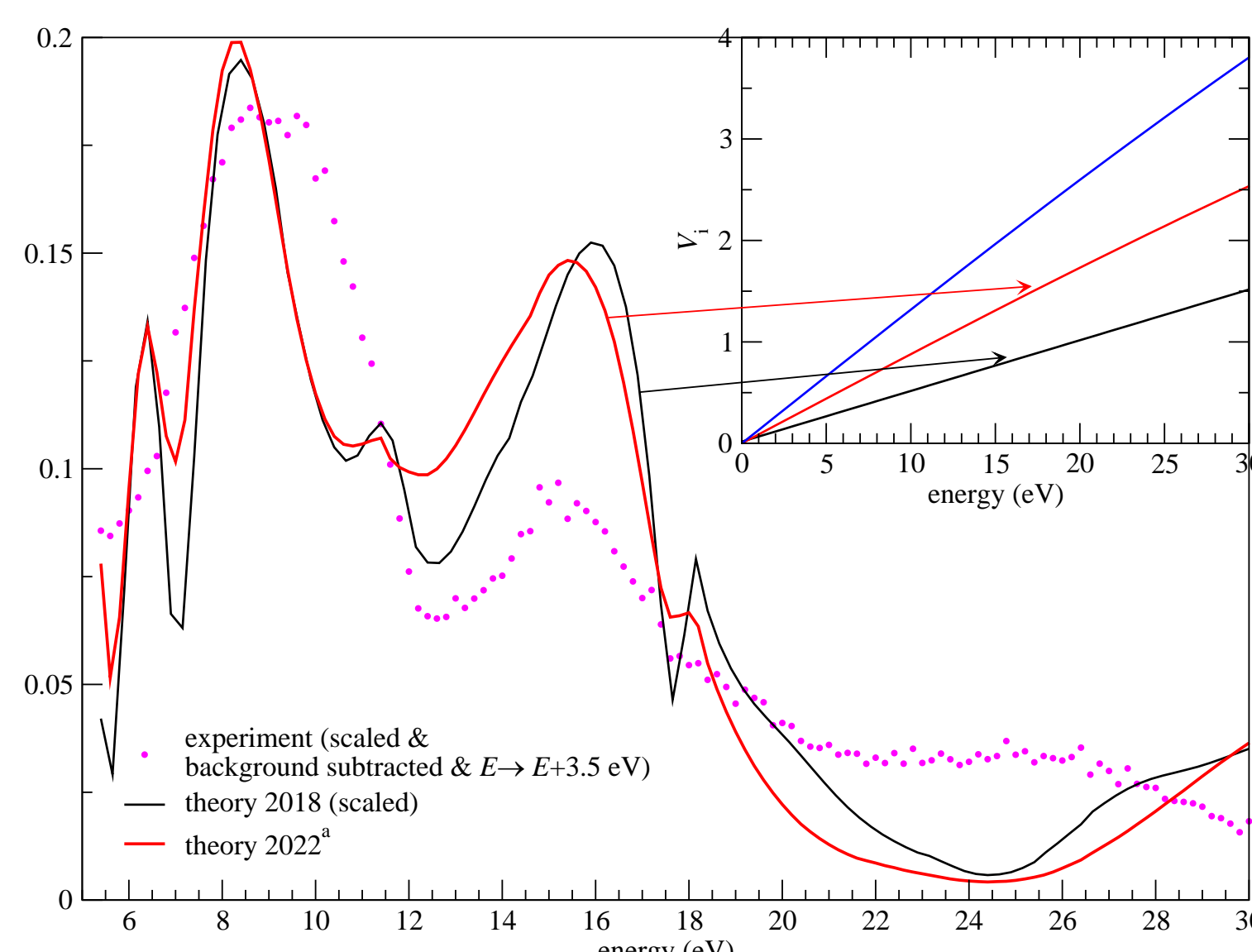
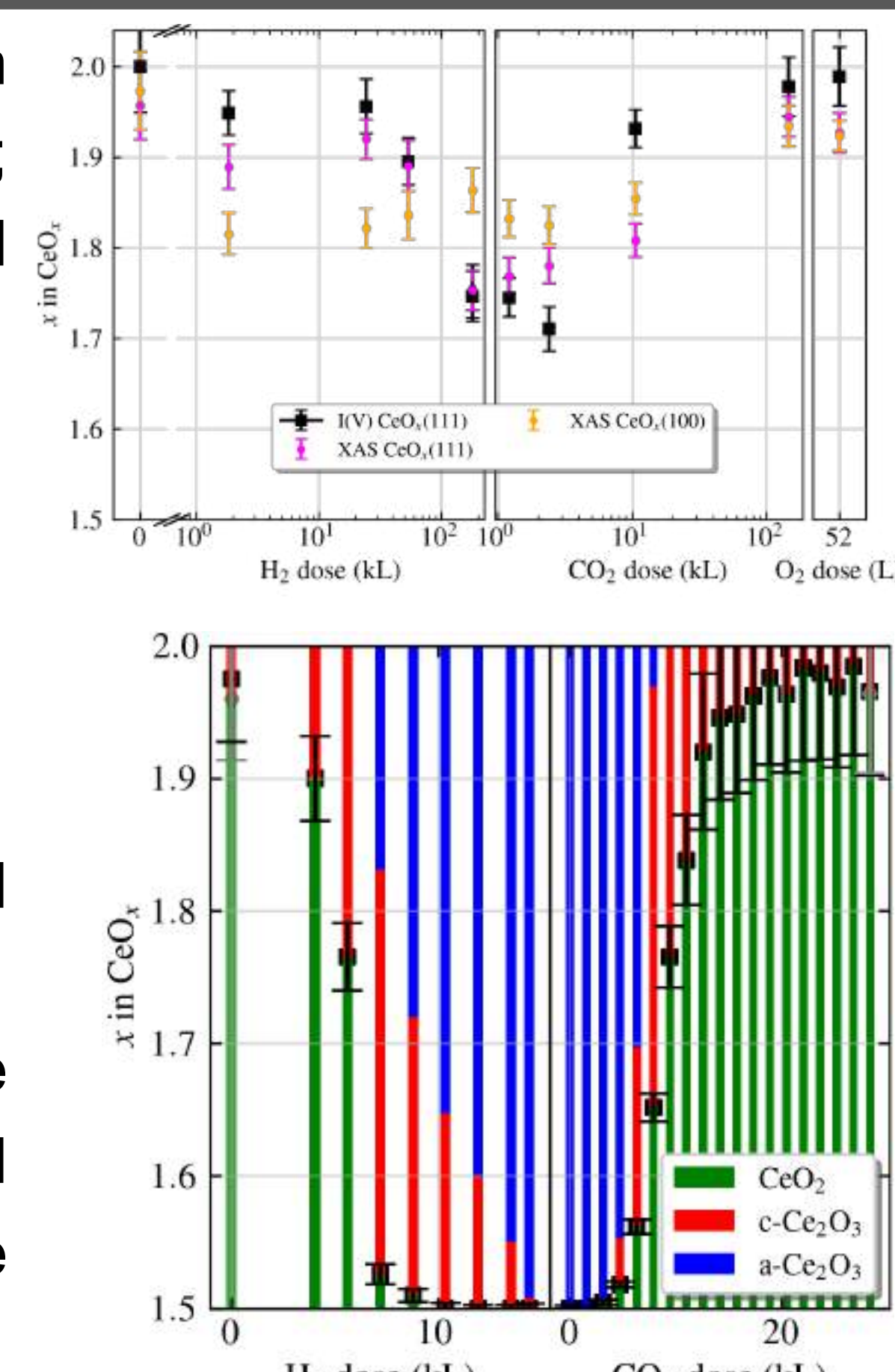


Figure 4. Dynamic I(V)-LEEM Reduction (left top), reoxidation (left bottom) and pixel-wise decomposition analysis [7-9] of the I(V) stacks in single images with increasing H₂ dose (0, 10.3 kL, 14.4 kL) and CO₂ dose (4.7 kL, 28.4 kL) (right). The triangle details the color map denoting the cerium cation oxidation state in the false-color images (right). (bottom right) First identification of hexagonal Ce₂O₃(0001) by comparison with ab initio scattering calculations [7, 10].



Summary and conclusions

- Reduction and reoxidation in agreement with literature [3,6]; stronger reduction achieved with higher H₂ dose.
- CeO_x(100) I(V) curve becomes featureless upon reduction and is not recoverable during reoxidation
- ⇒ loss of surface crystallinity
- Further redox cycles only need lower H₂ and CO₂ doses.
- CeO₂(111) reduces from the defects within the islands and reoxidizes starting from the island edges.



References and Acknowledgments

- [1] J. Graciani et al., Science **345**, 546 (2014).
- [2] P. M. Albrecht et al., Langmuir **29**, 4559 (2013).
- [3] J. Höcker et al., J. Phys. Chem. C **120**, 4895 (2016).
- [4] S. D. Senanayake et al., J. Phys. Chem. C **120**, 1778 (2016).
- [5] T. Duchon et al., J. Mater. Chem. A **8**, 5501 (2020).
- [6] T. Staudt et al., J. Catal. **275**, 181 (2010).
- [7] J. I. Flège et al., Phys. Rev. B **88**, 235428 (2013).
- [8] J. Höcker et al., Adv. Mater. Interfaces **2**, 1500314 (2015).
- [9] J. Höcker et al., Nanoscale **9**, 9352 (2017).
- [10] J. I. Flège & E. E. Krasovskii, Phys. Status Solidi RRL **8**, 463 (2014).



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